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EXAMINER				
KAPUSHOC, STEPHIN THOMAS				
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Please find below and/or attached an Office communication concerning this application or proceeding.

The time period for reply, if any, is set in the attached communication.

Office Action Summary

Application No.

10/824,656

Applicant(s)

FRANZEN ET AL.

Examiner

Stephen Kapushoc

Art Unit

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-- The MAILING DATE of this communication appears on the cover sheet with the correspondence address --
Period for Reply

A SHORTENED STATUTORY PERIOD FOR REPLY IS SET TO EXPIRE 3 MONTH(S) OR THIRTY (30) DAYS, WHICHEVER IS LONGER, FROM THE MAILING DATE OF THIS COMMUNICATION.

- Extensions of time may be available under the provisions of 37 CFR 1.136(a). In no event, however, may a reply be timely filed after SIX (6) MONTHS from the mailing date of this communication.
- If NO period for reply is specified above, the maximum statutory period will apply and will expire SIX (6) MONTHS from the mailing date of this communication.
- Failure to reply within the set or extended period for reply will, by statute, cause the application to become ABANDONED (35 U.S.C. § 133). Any reply received by the Office later than three months after the mailing date of this communication, even if timely filed, may reduce any earned patent term adjustment. See 37 CFR 1.704(b).

Status

- 1) ☒ Responsive to communication(s) filed on 17 January 2008.
- 2a) ☒ This action is **FINAL**. 2b) ☐ This action is non-final.
- 3) ☐ Since this application is in condition for allowance except for formal matters, prosecution as to the merits is closed in accordance with the practice under *Ex parte Quayle*, 1935 C.D. 11, 453 O.G. 213.

Disposition of Claims

- 4) ☒ Claim(s) 1, 4-8 and 11-19 is/are pending in the application.
- 4a) Of the above claim(s) _____ is/are withdrawn from consideration.
- 5) ☐ Claim(s) _____ is/are allowed.
- 6) ☒ Claim(s) 1, 4-8 and 11-19 is/are rejected.
- 7) ☐ Claim(s) _____ is/are objected to.
- 8) ☐ Claim(s) _____ are subject to restriction and/or election requirement.

Application Papers

- 9) ☐ The specification is objected to by the Examiner.
- 10) ☐ The drawing(s) filed on _____ is/are: a) ☐ accepted or b) ☐ objected to by the Examiner.
Applicant may not request that any objection to the drawing(s) be held in abeyance. See 37 CFR 1.85(a).
Replacement drawing sheet(s) including the correction is required if the drawing(s) is objected to. See 37 CFR 1.121(d).
- 11) ☐ The oath or declaration is objected to by the Examiner. Note the attached Office Action or form PTO-152.

Priority under 35 U.S.C. § 119

- 12) ☐ Acknowledgment is made of a claim for foreign priority under 35 U.S.C. § 119(a)-(d) or (f).
- a) ☐ All b) ☐ Some * c) ☐ None of:
1. ☐ Certified copies of the priority documents have been received.
 2. ☐ Certified copies of the priority documents have been received in Application No. _____.
 3. ☐ Copies of the certified copies of the priority documents have been received in this National Stage application from the International Bureau (PCT Rule 17.2(a)).

* See the attached detailed Office action for a list of the certified copies not received.

Attachment(s)

- 1) ☒ Notice of References Cited (PTO-892)
- 2) ☐ Notice of Draftsperson's Patent Drawing Review (PTO-948)
- 3) ☐ Information Disclosure Statement(s) (PTO/S5108)
Paper No(s)/Mail Date _____
- 4) ☐ Interview Summary (PTO-413)
Paper No(s)/Mail Date _____
- 5) ☐ Notice of Informal Patent Application
- 6) ☐ Other: _____

DETAILED ACTION

Claims 1, 4-8, and 11-19 pending and examined on the merits.

Please note: The text of those sections of Title 35, U.S. Code not included in this action can be found in a prior Office action.

This Office Action is in reply to Applicants' correspondence of 01/17/2008.

Applicants' remarks and amendments have been fully and carefully considered but are not found to be sufficient to put this application in condition for allowance. New grounds of rejection presented in this Office Action are necessitated by Applicants' amendments. Any rejections or objections not reiterated herein have been withdrawn in light of the amendments to the claims or as discussed in this Office Action.

This Action is **NON-FINAL**.

Claim Rejections - 35 USC § 112 1st ¶ - Written Description, New Matter

1. The following is a quotation of the first paragraph of 35 U.S.C. 112:

The specification shall contain a written description of the invention, and of the manner and process of making and using it, in such full, clear, concise, and exact terms as to enable any person skilled in the art to which it pertains, or with which it is most nearly connected, to make and use the same and shall set forth the best mode contemplated by the inventor of carrying out his invention.

2. Claims 1, 4-8, and 11-19 are rejected under 35 U.S.C. 112, first paragraph, as failing to comply with the written description requirement. The claim(s) contains subject matter which was not described in the specification in such a way as to reasonably convey to one skilled in the relevant art that the inventor(s), at the time the application was filed, had possession of the claimed invention. This is a **NEW MATTER** rejection.

Claim 1, from which 4-8 and 11-19 pending claims depend, have been amended to require, in step (f), 'measuring one of a substantially constant current and a substantially constant voltage'. This newly added limitation appears to limit the measurement of the current or voltage production from the required galvanic cell to require a specific portion of that production to be measured. It is noted that the

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specification as originally filed does not provide any teaching as to what is required for any current or voltage to be 'substantially constant'. As noted in the Remarks of 01/17/2008, page 7, the potential between the electrodes will drop when 'the electrode of the less-noble element or the electrolyte at the electrode of the more-noble element completely consumed'. Thus, as evidenced by Applicants' arguments, as well as the art cited in the instant Office Action, the current and voltage in a galvanic cell changes over time. The specification as originally filed does not specify what portion of galvanic cell current or voltage production is 'substantially constant'; nor does the specification indicate that a particular portion of any current or voltage produced will be measured.

As such, the specification as originally filed does not appear to support the required limitation of a step of 'measuring one of a substantially constant current and a substantially constant voltage generated between the electrodes of the galvanic cell'.

Withdrawn Claim Rejections - 35 USC § 103

3. The rejection of claims under 35 U.S.C. 103(a) as being unpatentable over Wang et al (2001) (as cited on the PTO-892 of 10/10/2006), as evidenced by Town (1997) and Wang (1994; US Patent 5,292,423) in view of Knoll (1999; WO 99/27367) are withdrawn in light of the amendments to the claims and Applicants' arguments.

Applicants have argued (p.6-7 of the Remarks) that the potentiometric stripping analysis (PSA) taught by Wang et al does not provide a method wherein the working electrode forms an electrode of a galvanic cell (as required by claim 1) where a substantially constant current or voltage is generated and measured. It is noted that

new rejections, incorporating the teachings of Jaya et al (1985), are set forth in this Office Action that address the limitations of the claims.

New Claim Rejections - 35 USC § 103

4. Claims 1, 4, 7, 8, and 11 are rejected under 35 U.S.C. 103(a) as being unpatentable over Wang et al (2001) (as cited on the PTO-892 of 10/10/2006) in view of Jaya et al (1985) and Knoll (1999; WO 99/27367).

Wang et al teaches a method for measuring the binding of a nucleic acid analyte molecule to a probe molecule.

Regarding claim 1, Wang et al teaches providing a circuit surface having a contact spot and a metal counterelectrode (Fig 1; p.5577 – Electrode preparation), relevant to part (a). Relevant to part (b) of claim 1, Wang et al teaches the immobilization of probe molecules (Fig 1; p.5777 – Preparation of oligomers coated microspheres and analytical procedure). Relevant to parts (c) and (d), Wang et al teaches binding streptavidin coated gold particles to the analyte molecules, and binding of the analyte molecules to the probe molecules (Fig 1; p.5578, left col., Ins.5-11). Relevant to step (e), Wang et al teaches a counterelectrode that is Ag/AgCl and a nanoparticle surface that is gold, which are metals from an electrochemical series, and teaches analysis of the gold metal from the nanoparticle using potentiometer stripping analysis (PSA), where in PSA the gold is deposited on the contact spot of the circuit (which establishes electrical contact) in the presence of an electrolyte. Relevant to step (f), Wang et al teaches measuring electrical properties of the circuit created upon metal

binding (Fig 3) and teaches that measurement allows the binding of the analyte molecule to the probe to be measured (p.5579 – Analytical performance).

Regarding claim 8, Wang et al teaches that the gold nanoparticles are bound to the analyte after the analyte is bound to the capture probe (p.5578, left col., lns.8-11)

Wang et al does not specifically teach detection methods comprising measuring a substantially constant current or voltage generated between the electrodes of a galvanic cell, as required by step (f) of claim 1.

However, such measurements in a galvanic cell for the detection of metals, where Wang et al teaches a metal as part of a nanoparticle label of a nucleic acid analyte, were well known at the time the invention was made.

Jaya et al teaches a method, referred to as galvanic stripping analysis (GSA), where the presence of a metal analyte on a working electrode is detected as a galvanic cell is formed between the metal analyte and a platinum foil counter electrode (p.1443, lns.10-18; Fig. 1c; p.1450, lns. 15-18). Relevant to the required limitations of claim 1, Jaya et al teaches measuring a substantially constant voltage generated between the electrodes of the galvanic cell (p.1447, lns.1-6; Figs.2-4), where Jaya et al teaches measuring a range of voltage from the galvanic cell which includes a substantially constant voltage.

Neither Wang et al nor Jaya et al teaches probes immobilized in spatial proximity to the electronic circuits (as required by part (b) of claim 1), or probes immobilized on the circuit surface (claim 4) or a countersurface (claim 5). Neither Wang et al nor Jaya et al teaches a method in which nanoparticles are bound to analytes before the analytes

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are hybridized to the capture probe (claim 7), or methods where the electrical contact between the nanoparticle and the contact spot is established by electrically conductive molecules (claim 11).

However, such methods for electrically-based nucleic acid detection were well known in the art at the time the invention was made.

Knoll et al teaches a method for electronic detection of the binding of analyte molecules to probe molecules. Relevant to step (b) of claim 1, Knoll teaches providing immobilized probe molecules in spatial proximity to a circuit surface having electronic circuits (see for example Fig 9 and 10).

Regarding claim 4, Knoll teaches probe molecules bound to the circuit surface (for example Fig 10).

Regarding claim 7, Knoll teaches the marker particle bound to the analyte prior to binding of an analyte to a probe (Fig 9), relevant to claim 7.

Relevant to claim 11, Knoll teaches that the marker particles may be conductive (col.14 Ins.5-9), thus they are electrically conductive molecules.

It would have been prima facie obvious to one of ordinary skill in the art at the time the invention was made to have modified the methods of Wang et al, which use PSA techniques to measure a metal label for detection of nucleic acid binding, by using the GSA technique to detect a metal label. One would have been motivated to use the GSA methods of Jaya et al based on the teaching of Jaya et al that GSA is a rapid, simple, and versatile technique (p.1441 – Abstract) with advantages over other techniques (p.1454).

Furthermore, it would have been prima facie obvious to one of ordinary skill in the art at the time the invention was made to have used the techniques of Knoll in the method for measuring the binding of analytes to probes as taught by Wang et al in view of Jaya et al. One would have been motivated to use the methods of Knoll et al because Knoll et al teaches the successful deposition of conductive elements (termed 'marker particles by Knoll) using probes immobilized in spatial proximity to an electronic circuit of interest, and both Wang et al (Fig 1) and Jaya et al (Fig 1) teach that such deposition is required for the sensitive detection of analyte binding by using electrical methods including GSA (Jaya et al p.1448-1449).

Response to remarks

5. It is noted Applicants' remarks concerning the deficiencies of Wang et al with regard to the limitations of 'measuring one of a substantially constant current and a substantially constant voltage generated between the electrodes of the galvanic cell' have been addressed previously in this Office Action. Applicants' arguments are moot in view of the new rejection set forth in this Office Action which cites the teachings of Jaya et al, which provides a method of detecting a metal label using the required measurements.

Applicants have further argued (p.8 of Remarks) that, with regard to claim 11, that the teachings of Knoll et al do not include additional electrically conductive molecules as recited in claim 11. This argument has been fully and carefully considered but is not found to be persuasive. The limitations of claim 11 require that the contact between the nanoparticle the contact spot is established by electrically conductive

molecules. There is no requirement in the claim that the nanoparticles can not comprise the conductive molecules. As such Applicant's are arguing limitations that are not required by the claims when arguing that Knoll et al does not teach electrically conductive molecules that are some how specifically additional the nanoparticles. In the case of Knoll, where the nanoparticles have electrically conductive molecules, the teachings of Knoll satisfy the limitations of the claims.

6. Claims 6, 18 and 19 are rejected under 35 U.S.C. 103(a) as being unpatentable over Wang et al (2001) (as cited on the PTO-892 of 10/10/2006) in view of Jaya et al (1985) and Knoll (1999; WO 99/27367) as applied to claims 1, 4, 7, 8, and 11 above, and further in view of Henkens et al (2002, US Patent 6,391,558).

The teachings of Wang et al in view of Jaya et al and Knoll are applied to claims 6, 18, and 19 as they were previously applied to claims 1, 4, 7, 8, and 11.

Regarding claim 6, Knoll teaches probe molecules immobilized to an electrode surface, and analyte molecules affinity bound (e.g. antibody-antigen, Fig 5; DNA:DNA / probe:analyte hybridization) to the probe molecules, and Wang et al teaches biotin:avidin immobilization of a probe.

Wang et al in view of Jaya et al and Knoll does not particularly require that the immobilized probe is bound by a covalent bond (claim 6), or provide any particular details regarding the binding of the nanoparticle to the analyte molecule (claims 18 and 19).

Henkens et al teaches methods for the detection of nucleic acids using electrodes comprising immobilized probes, as well as analyte molecules labeled with detectable reporters.

Regarding claim 6 Henkens et al specifically teaches that capture probes may be covalently bound to an electrode (col.45 Ins.16-25).

Regarding claims 18 and 19, Henkens et al teaches the PCR amplification of a analyte DNA molecule using primers modified at the 5' end, and gives the examples of fluorescein and biotin labeled primers (col.21 ln.60 - col 22. ln.4). Henkens et al indicates that the resulting labeled PCR product may be attached to a reporter molecule by an interaction between the label from the PCR primer and a binding partner for the label of the primer. Relevant to claim 18, Henkens et al teaches the biotin:avidin binding pair, as well as labeling a PCR product using a biotinylated primer, and binding of the labeled PCR product to a avidin-gold binding partner (for example col.5 Ins.25-35, col.43 ln.55). Relevant to claim 19, Henkens et al particularly teaches binding of a fluorescein-labeled PCR product to an anti-fluorescein HRP conjugate.

It would have been prima facie obvious to one of ordinary skill in the art at the time the invention was made to have used the nucleic acid probe immobilization methods of Henkens et al in the electrode based analysis methods of Wang et al in view of Jaya et al and Knoll. One would have been motivated to do so based on the assertion of Henkens et al that covalent attachment of a probe is a preferred method (col.45 Ins. 16-25). It would further have been obvious to use the probe:reporter binding method of Henkens et al to accomplish the marker particle:analyte binding of Wang et al

in view of Knoll et al. One would have been motivated to do so because Henkens et al teaches that such methods can be used to attach a variety of different molecules (including colloidal gold which is similar to the description of marker particles by Knoll) to nucleic acid for analysis.

Response to Remarks

Applicants argue (p.8 of Remarks), that Henkens does not disclose the formation of a galvanic cell that, in turn, generates output. However, the limitations of the claims which require formation of a galvanic cell are taught by Jaya et al, as addressed previously in this Office Action. In the rejection, Henkens is not relied upon for any teachings of galvanic cell formation.

7. Claims 5 and 12 are rejected under 35 U.S.C. 103(a) as being unpatentable over Wang et al (2001) (as cited on the PTO-892 of 10/10/2006) in view of Jaya et al (1985) Knoll (1999; WO 99/27367), as applied to claims 1, 4, 7, 8, and 11 above, and further in view of Wohlstadter et al (2001, US Patent 6,207,369).

As discussed in detail in the rejection of claims 1, 4, 7, 8, and 11 earlier in this Office Action, Wang et al in view of Jaya et al and Knoll teaches all of the required limitations of claims 1 and 11, from which claims 5 and 12 depend.

Wang et al in view of Jaya et al and Knoll does not specifically teach probe molecules bound to a countersurface positioned opposite the circuit surface (claim 5), or the use of polyene molecules to conduct an electrical signal (claim 12).

Regarding claim 5, Wohlstadter et al teaches methods of using several configurations of electrode-based devices in which the portion of the device where the analyte is collected (termed in the reference the 'binding domain') is on a surface opposite from an electrode (see for example Fig. 21 and Fig. 37).

Regarding claim 12, Wohlstadter et al teaches the use of a linking chain to ensure low resistance of electron transfer from the electrode, and specifically teaches the use of a polyacetylene chain (col.39 lns.53-63), which is of the polyene class.

It would have been prima facie obvious to one of ordinary skill in the art at the time the invention was made to have performed the electronic detection methods of Wang et al in view of Jaya et al and Knoll using a probe binding area on a countersurface opposing an electrode. One would have been motivated to do so based on the teachings of Wohlstadter et al that in such a configuration the electrode can be protected during the binding reaction from the sample by a physical barrier that is subsequently removed thus, preventing contamination of the electrode surface which could result in a change in its electrochemical performance (col.64 lns.1-11). One would have a reasonable expectation of success because Wohlstadter et al teaches that the binding domain of the countersurface makes contact with the electrode and carries current from the counter electrode to the working electrode (col.45 lns.9-35). It would have been further obvious to use the polyacetylene chains of Wohlstadter et al to ensure low resistance of conductivity from the electrode to the marker particle as Wohlstadter et al teaches this use for polyacetylene chains.

Response to Remarks

Applicants argue (p.9 of Remarks), that Wohlstadter et al does not disclose forming a galvanic cell and the measuring the electrical properties of that element as recited. However, the limitations of the claims which require formation of a galvanic cell are taught by Jaya et al, as addressed previously in this Office Action. In the rejection, Wohlstadter et al is not relied upon for any teachings of galvanic cell formation.

8. Claims 13, 16, and 17 are rejected under 35 U.S.C. 103(a) as being unpatentable over Wang et al (2001) (as cited on the PTO-892 of 10/10/2006) in view of Jaya et al (1985) and Knoll (1999; WO 99/27367), as applied to claims 1,4, 7, 8, and 11 above, and further in view of Fish (2002, WO 02/054052 A1).

As discussed in detail in the rejection of claims 1, 4, 7, 8, and 11 earlier in this Office Action, Wang et al in view of Jaya et al and Knoll teaches all of the required limitations of Claim 1, from which claims 13, 16, and 17 depend.

Relevant to claim 16, Knoll specifically teaches that the marker particles may be magnetic (see for example col.4 lns.10-24).

Relevant to claim 17, Knoll teaches that the marker particles may be dendrimers (col.14 lns.2-3), which are protrusions.

Wang et al in view of Jaya et al and Knoll does specifically teach a requirement of nanoparticles touching a contact spot, though Wang et al and Jaya et al both teach the requirement of the metal label contacting an electrode.

Fish teaches the detection of analytes using an electrode-based method wherein an opposing surface with an electrode is moved to make contact with an electrically

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readable particle that is bound to and analyte, where the analyte is bound to an immobilized probe (see for example Fig 1, p.14-18). Regarding claim 13, Fish specifically teaches that pressure is applied to the particle (p.16 last two lines) and that the bound particles make contact with the electrode (p.17 lns.7-8).

Regarding claim 16, Fish teaches that a countersurface may be moved in order to create a physical contact between an electrode (which is a contact spot) and an electrically readable particle, and also teaches that the probes may be attached to a contact spot and that movement of the countersurface causes the particles to make contact with the contact spot (see for example Fig 13 and page 28).

Additionally relevant to claim 17, Fish teaches that an electrode may be rough and have sharp edges and vertices to make electrical contact (p.42), thus teaching a circuit surface with electrically conductive protrusions.

It would have been *prima facie* obvious to one of ordinary skill in the art at the time the invention was made to have performed the electrode-based analyte detection method of Wang et al in view of Knoll by incorporating the countersurface movement taught by Fish to make contact between a particle and an electrode. One would have been motivated to do so based on the teachings of Fish that such methods allow accurate electrochemistry to be performed quickly at a low cost (p.7), and the teaching of the methods of Wang et al requiring contact between the element of the nanoparticle and the electrode. One would have had a reasonable expectation of success because Wang et al in view of Knoll teaches that the electrode-based method can be used as a

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multi-step process separating the steps of particle transport and electrode binding (Knoll et al at col. 5 lns.62-67).

Response to Remarks

Applicants argue (p.9 of Remarks), that Fish detects the presence and quantity of the analyte molecules by measuring electrical changes in a cell, and does not disclose forming a galvanic cell and the measuring the electrical properties of that element as recited. However, the limitations of the claims which require formation of a galvanic cell are taught by Jaya et al, as addressed previously in this Office Action. In the rejection, Fish is not relied upon for any teachings of galvanic cell formation

9. Claims 14 and 15 are rejected under 35 U.S.C. 103(a) as being unpatentable over Wang et al (2001) (as cited on the PTO-892 of 10/10/2006) in view of Jaya et al (1985) and Knoll (1999; WO 99/27367) and Fish (2002, WO 02/054052 A1), as applied to claims 13, 16, and 17 above, and further in view of Wohlstadter et al (2001, US Patent 6,207,369).

The teachings of Wang et al in view of Jaya et al, Knoll, and Fish are applied to claims 14 and 15 as they were previously applied to claims 13, 16, and 17.

Wang et al in view of Jaya et al, Knoll, and Fish teaches an electrode-based method of analyte detection wherein marker particles bind to analyte molecules and contact between the marker particle and a contact spot is made by the nanoparticles touching the contact spot.

Regarding claim 15, Wang et al a method in which a gold nanoparticle binds to a target oligonucleotide wherein the target oligonucleotide has hybridized to a probe oligonucleotide immobilized to a solid support (Fig 1; p.5577, left col., Ins.5-10). The method of Wang et al includes a step of dissolution of the gold nanoparticle from the analyte molecule prior to detection of the gold nanoparticle at an electrode.

Wang et al in view of Jaya et al, Knoll, and Fish does not specifically teach that the analyte molecule:particle complex is located on a surface opposite the circuit surface.

Wohlstadter et al teaches methods of using several configurations of electrode-based devices in which the portion of the device where the analyte is collected (termed in the reference the 'binding domain') is on a surface opposite from an electrode (see for example Fig. 21 and Fig. 37).

It would have been prima facie obvious to one of ordinary skill in the art at the time the invention was made to have used the opposed binding and electrode surfaces taught by Wohlstader et al in the electrode-based analyte detection method of Wang et al in view of Jaya et al, Knoll, and Fish. One would have been motivated to do so based on the teachings of Wohlstadter et al that in such a configuration the electrode can be protected during the binding reaction from the sample by a physical barrier that is subsequently removed thus, preventing contamination of the electrode surface which could result in a change in its electrochemical performance (col.64 Ins.1-11).

Regarding the limitations of claim 15, It would have been prima facie obvious to one of ordinary skill in the art at the time the invention was made to have used the

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dissolution of a gold microparticle from a bound analyte molecule as taught by Wang et al as a detectable nanoparticle in the electrode-based analyte detection method of Knoll in view Fish and Wohlstadter et al. One would have been motivated to use such a method because Wang et al teaches the sensitivity ((p.5581, left col., Ins.13-16) and adaptability (p.5581, right col., Ins.17-19) of such a technique. Incorporating the methods of Wang et al into the teachings of Knoll et al would result in the use of a magnetic field to move an electrically conductive and magnetic marker particle to an electrode after separation of the particle from a particle:analyte:probe complex.

Response to Remarks

Applicants' arguments (p.10 of Remarks) concerning the deficiency of the references with regard to the teaching of formation of a galvanic cell have been addressed previously in this Office Action.

Pertinent Prior Art

10. The prior art made of record and not relied upon is considered pertinent to applicant's disclosure. Jaya et al (July 1987) in Analyst and Jaya et al (December 1987) in Analyst teach the detection of the metals lead and cadmium using the galvanic stripping analysis (GSA) methods.

Conclusion

11. No claim is allowable.

Applicant's amendment necessitated the new ground(s) of rejection presented in this Office action. Accordingly, **THIS ACTION IS MADE FINAL**. See MPEP § 706.07(a). Applicant is reminded of the extension of time policy as set forth in 37 CFR 1.136(a).

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A shortened statutory period for reply to this final action is set to expire THREE MONTHS from the mailing date of this action. In the event a first reply is filed within TWO MONTHS of the mailing date of this final action and the advisory action is not mailed until after the end of the THREE-MONTH shortened statutory period, then the shortened statutory period will expire on the date the advisory action is mailed, and any extension fee pursuant to 37 CFR 1.136(a) will be calculated from the mailing date of the advisory action. In no event, however, will the statutory period for reply expire later than SIX MONTHS from the date of this final action.

Any inquiry concerning this communication or earlier communications from the examiner should be directed to Stephen Kapushoc whose telephone number is 571-272-3312. The examiner can normally be reached on Monday through Friday, from 8am until 5pm.

If attempts to reach the examiner by telephone are unsuccessful, the examiner's supervisor, Ram Shukla can be reached at 571-272-0735. The fax phone number for the organization where this application or proceeding is assigned is 571-273-8300.

Information regarding the status of an application may be obtained from the Patent Application Information Retrieval (PAIR) system. Status information for published applications may be obtained from either Private PAIR or Public PAIR. Status information for unpublished applications is available through Private PAIR only. For more information about the PAIR system, see <http://pair-direct.uspto.gov>. Should you have questions on access to the Private PAIR system, contact the Electronic Business Center (EBC) at 866-217-9197 (toll-free).

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/Stephen Kapushoc/
Art Unit 1634

/Jehanne S Sitton/
Primary Examiner, Art Unit 1634